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$- KE_i = -7320 \text{ J}$

$\Delta U_{\text{hands}} = (1 -$

$0.300)\Delta U \quad \Delta PE$

$+ \Delta KE + \Delta U = 0$ The

hands don't

change height,

so $\Delta PE = 0 \text{ J}$. ΔU

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water is heated

from 23 degrees

C to 78 degrees

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C. What is the change in its temperature on the Kelvin and Fahrenheit scales? A temperature difference of 1 degree in Celsius is equal to a temperature difference of 1 degree in Kelvin. Celsius and Kelvin differ

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