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How are equations used to represent chemical reactions? - OCR 21C test questions - OCR 21st Century

Salt A is sodium chloride, NaCl. The yellow flame colour shows that Na⁺ ions are present, and the white silver halide precipitate shows that Cl⁻ ions are present. Suggest an explanation for a ...

Core practical

Here, they lose electrons to form chlorine atoms. The atoms join up in pairs to form Cl₂ molecules, so chlorine gas is formed at the positive electrode. During the electrolysis of molten salts ...

Electrolysis of molten salts

However, a fluoride ion is not a neon atom. The nucleus of a fluoride ion is the nucleus of a fluorine atom and has 9 protons - but the nucleus of a neon atom has 10.

Representing negative ions

The Group 7 elements - fluorine (F), chlorine (Cl), bromine (Br), iodine (I) and astatine (At) - have seven electrons in the outer shell.

Atomic number, mass number and isotopes

However, we also know not everyone is into horror, so don't worry, this month also brings literary fiction titles from heavy hitters like Jonathan Franzen, romance reads that will leave you ...

The 55 Best New Books of October Are the Perfect Mix of Horror, Romance, and Nonfiction

Look closely at this image, stripped of its caption, and join the moderated conversation about what you and other students see. By The Learning Network In these three short lesson plans based on ...

The Learning Network

For example, Group 1 contains very reactive metals such as sodium (Na), while Group 7 contains very reactive non-metals such as chlorine (Cl). There are no compounds in the periodic table ...

The periodic table

Literary sisters Charlotte Emily and Anne Bronte lived in the parsonage in Haworth with their brother Branwell, and between them wrote Jane Eyre, Wuthering Heights and The Tenant Of Wildfell Hall.

In Pictures: Like clockwork, steampunk fans descend on Bronte village

What is produced in the backward reaction in this equilibrium, when the reaction goes from right to left: NH₄Cl(s) ? NH₃(g) + HCl(g)?

Reversible reactions, ammonia and the Haber process test questions - WJEC

According to the BBC, fewer than half of players at most Premier ... The way they won the 2011 CL and 2015 was mesmerizing. In addition to that we seen one of the greatest national sides of ...

There is one manager for Man Utd on a par with Pep and Klopp

However, a fluoride ion is not a neon atom. The nucleus of a fluoride ion is the nucleus of a fluorine atom and has 9 protons - but the nucleus of a neon atom has 10. Oxygen is in Group 6.

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