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produced by hydrogen gas. The link between light and atomic structure is illustrated by the Bohr Model of Hydrogen Gizmo. The Gizmo shows a laser pointed at a container of hydrogen gas. After

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passing through the gas, light from the laser goes through a prism and is detected on a screen.

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KEY)

Shoot a stream of photons through a container of hydrogen

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gas. Observe how photons of certain energies are absorbed, causing the electron to move to different orbits. Build the spectrum of hydrogen based on photons that are absorbed and emitted.

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Hydrogen Shoot a

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hydrogen gas.

Observe how photons
of certain energies
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the electron to move
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photons that are
absorbed and
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1913, Niels Bohr

proposed that the

unique spectral lines

created by different

elements were related

to the way electrons

were arranged around

the nucleus. The Bohr

Model:...

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Introduction

(ANSWER KEY ...

The Bohr model of the atom was inspired by the spectrum

produced by

hydrogen gas. The

link between light and

atomic structure is

illustrated by the Bohr

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shows a laser pointed

at a hydrogen atom.

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at a container of hydrogen gas. After passing through the gas, light from the laser goes through a prism and is detected on a screen.

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gas. Observe how

photons of certain

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absorbed, causing the

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Key Vocabulary:
absorption spectrum,
Bohr model, electron
volt, energy level,

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laser, orbital, photon

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Questions (Do these

BEFORE using the

Gizmo.) [Note: The

purpose of these

questions is to

activate prior

knowledge and get

students thinking.

Students are not

expected to know the

answers to the Prior

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