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reaction proceeds in an  
Anti-Markovnikov  
manner, where the  
hydrogen (from  $\text{BH}_3$  or  
 $\text{BHR}_2$ ) attaches to the  
more substituted carbon  
and the boron attaches  
to the least substituted  
carbon in the alkene  
double bond.

Furthermore, the borane  
acts as a Lewis  
Anti-  
Markovnikov acid by  
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