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(a) A dative covalent or coordinate covalent bond is one in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written

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A?B, where the direction of the arrow shows the direction in which the electron pair is donated.

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(a) (i) Define the term bond
energy [1. mark] The bond
energy represents: The
quantity of energy evolved
when the bond is formed.

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Also, the bond energy is the amount of energy absorbed when the bond is broken.

(ii) The bond lengths and bond energies of carbon-carbon single, double and triple bonds are listed below.

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