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## Chapter 8 Chemistry Of The Main Group Elements

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## **Chemistry Of The**

Chapter 8 Chemistry of the Main Group Elements 107 Copyright © 2014 Pearson Education, Inc. CHAPTER 8: CHEMISTRY OF THE MAIN GROUP ELEMENTS 8.1 a.  $H_2$  74.2 pm 436 kJ/mol  $H_2 + 106$  pm 255 kJ/mol These values are consistent with the molecular orbital descriptions.  $H_2$  has two electrons in the bonding orbital,  $H_2 +$  has only one. The net attraction between the

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## **CHAPTER 8: CHEMISTRY OF THE MAIN GROUP ELEMENTS**

Share –(Long Questions)–[Open] ???????????? –(Test Yourself / Conceptual Questions)–[Open] ?????????? –(Exercise: MCQ's+Short Q+Long Q)–[Open] ?????????? –(Extra: MCQ's)–[Open] ?????????????? PAST PAPERS SHORT QUESTIONS LAHORE & GUJRANWALA What is electropositivity? Explain with an example. LHR 2014 ...

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sharp knife will exert a higher pressure with the same amount of force and cut through material more effectively. 3.

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Home > GCSE study tools > Chemistry > (Chapter 8) Chemical Analysis (Chapter 8) Chemical Analysis. 0.0 / 5. Hide Show resource information. Chemistry; Testing and analysing substances Pure Substances Formulations; GCSE; AQA; Created by: 14randalloli; Created on: 05-12-17 18:06; What is a pure substance?

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Explain why ? Answer: Zinc ( $Z=30$ ) has the configuration  $3d^{10} 4s^2$ . It does not have partially filled d-subshell in its elementary form or in commonly occurring oxidation state ( $Zn^{2+} = 3d^{10}$ ). Therefore, it is not regarded ...

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Figure 8.8.1 Schematic Diagram of a Wet Scrubber System. In coal-burning power plants,  $\text{SO}_2$  can be removed (“scrubbed”) from exhaust gases by its reaction with a lime ( $\text{CaO}$ ) and water spray to produce calcium sulfite dihydrate ( $\text{CaSO}_3 \cdot 2\text{H}_2\text{O}$ ). Removing  $\text{SO}_2$  from the gases prevents its conversion to  $\text{SO}_3$  and subsequent reaction with rainwater (acid rain). Scrubbing systems are now commonly used to minimize the environmental effects of large-scale fossil fuel combustion.

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## **Chapter 8 Chemistry Of The Main Group Elements**

Active Themes Upstairs, the Time Traveller finds a gallery of technical chemistry where there is a well-preserved box of matches under glass, as well as a jar of camphor. Finding these light sources gives the Time Traveller hope for retrieving his time machine, though he passes the rest of the day without finding any more useful implements.

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Both types of bonds result from overlap of atomic orbitals on adjacent atoms and contain a maximum of two electrons.

Differences:  $\sigma$  bonds are stronger and result from end-to-end overlap and all single bonds are  $\sigma$  bonds;  $\pi$  bonds between the same two atoms are weaker because they result from side-by-side overlap, and multiple bonds contain one or more  $\pi$  bonds (in addition to a  $\sigma$  bond).

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