

## Chapter 8 Covalent Bonding Workbook Answers

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Chapter 8 Covalent Bonding 183 Section Review Objectives

- State a rule that usually tells how many electrons are shared to form a covalent bond
- Describe how electron dot formulas are used
- Predict when two atoms are likely to be joined by a double or a triple covalent bond
- Distinguish between a single covalent bond and other covalent bonds

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- Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in ...

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electrons repel because they are both negatively charged.

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Chapter 8 Covalent Bonding. chemical bond. covalent bond. molecule. electron dot diagram. 化学键 the force that holds two atoms together. 共价键 the chemical bond that results from sharing valence electro.... 分子 a molecule is formed when two or more atoms bond covalently. 电子点式 used to show valence electrons of atoms.

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