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the physical state in which the opposing process of dissolution and crystallization of a solute occur at equal rates. Saturated solution. a solution that contains the maximum amount of dissolved solute. Unsaturated solution. a solution that contains less solute than a saturated solution under the existing conditions.

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Chemistry | Chapter 12 | Solutions. the ability of one substance to dissolve in another at a given temperature and pressure; expressed in terms of the amount of solute that will dissolve in a given amount of solvent to produce a saturated solution. the law that states that at constant temperature, the solubility of a gas in a liquid is directly proportional to the partial pressure of the gas on the surface of the liquid.

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