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IMPORTANT
QUESTIONS,
REPEATED~~

*QUESTIONS Metals
And Non-Metals
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|Faraday's Law
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Atoms and
Molecules in 30
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~~Stoichiometry
and Calculations
with Formulas
and Equations:~~

~~Part 1 of 5~~

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~~Stoichiometry,
Formulas and~~

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Equations: Part

3 of 8 *Four*

Chemical

Reactions:

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2 An Intro to

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Reactions:

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1 How to Balance

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& Molecules
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Equations: Part

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(Atoms,

Molecules and

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\u0026 Chapter 3

(Stoichiometry)

- Part 1 Atoms

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Review. a system
of physical
units (SI Units
) based on the
meter, kilogram,
second, ampere,
kelvin, candela,
and mole,
together with a
set of prefixes
to indicate
multiplication

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or division by a
power of ten.

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all the
questions in

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sentences (just restate the original

question). Show the math work

for # 27-30. 1.

The law of conservation of mass follows

from the concept that _____ . 2.

If 4 g of element A

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combine with 10
g of element B,
then 12 g of
element A
combine with
_____ g of
element B. 3.

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Chapter 3

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Review. dalton's

five principles
of the atomic...

law of

conservation of
mass. law of

multiple

proportions. law
of definite

proportions. 1.

all matter is

composed of

extremely small

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particles called... mass is
neither created
nor destroyed in
a chemical
reaction b...

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that has mass
and takes up
space. matter
composed of all
one type of atom
(can be a
single element,
like He or a
diatomic
element, like O₂
or H₂). Elements
cannot be broken
down into
simpler

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chemical
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combinations of
two or more
elements.

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The presentation of information in this chapter assumes that you can already perform the tasks listed below. You can test your readiness to proceed by answering the Review Questions at the

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end of the chapter. This might also be a good time to read the Chapter . Objectives, which precede the Review Questions.

Chapter 3 - An Introduction to Chemistry: Chemical

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authors of guide
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can discover
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up of motionless
particles.

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Matter is made
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of conservation
of mass. mass is
neither created
nor destroyed
during ordinary
chemical

reactions or
physical changes
 $H_2 + O = H_2 O$. law
of definite
proportions.

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compounds are
made of the same
elements in the
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proportions. law

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measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called _____.

17. The energy required to remove one

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electron from an
atom is called
its _____. 18.

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