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Pure Substances and

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Examples,

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the differences

between elements,

compounds and

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5th.. Matter And

A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A change in chemical properties of a substance like composition,

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reactivity with water, corrosiveness, and pH is known as chemical change. In chemical change, a new substance will be formed.

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Chapter 9 145

CHAPTER 9

SOLUTIONS MANUAL

35. Aqueous

solutions of

potassium iodide and

silver nitrate are

mixed, forming the

precipitate silver

iodide. chemical

equation: $KI(aq)$

$AgNO_3(aq) \rightarrow KNO_3(aq)$

$AgI(s)$ complete

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ionic equation: $K(aq) + I(aq) + 3AgNO_3(aq) \rightarrow 3AgI(s) + 3KNO_3(aq)$

Manual Chapter

Chemical

Reactions

Reactions

Solution. Physical:

boiling and melting are physical changes.

When water boils, no bonds are broken or formed. The change

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could be written:



Chemical:

The dark grey nail changes color to form an orange flaky substance (the rust); this must be a chemical change.

3.6: Changes in
Matter: Physical and

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Chemical Changes ...

240 Chemistry:

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SOLUTIONS MANUAL

30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the

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substance goes from
the solid phase
directly to the vapor
phase. During
evaporation, particles
in a liquid gain
enough

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