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Part 1 (1-25) 1) When a

beam of white rays is

dispersed by a prism

which colour will be

refracted to a larger

extent? Answer: Violet.

2) The modern atomic

mass unit is based on

\_\_\_\_\_ Answer:

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Carbon-12. 3) What do you mean by photoelectric effect?

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Question & Answers (0)

The relationship between the coefficients of the components of a balanced chemical is referred to as the: a) tracheotomy b)

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symmetry c)  
stoichiometry d)  
numerology e )...

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This hub examines the answers to some of the top chemistry-related science questions I have been asked by my students in our science lessons. 1. What Is an

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Acid? Put simply, an acid is any substance that has pH of less than 7. The pH scale is used to measure how acid or alkali a substance is: 0-3 = strong acid (UI turns red)

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1. What is the first  
element on the periodic  
table? 2. What is the

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centre of an atom  
called? 3. True or false?  
Acids have a pH level  
below 7. 4. What is the  
main gas found in the  
air we breathe?

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had been a million  
dollars on the line for  
knowing all the  
answers, you would  
have just missed the  
prize. It's OK to miss a

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question or two. That just shows you have other interests! You have a good command of chemistry.

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Answers: Nickel, cobalt  
and gadolinium.

Helium; Xenon; Sir  
Humphry Davy; Van

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der Waals forces (or  
bonds) (a)Antimony,  
(b)Potassium, (c)Silver,  
(d)Tungsten, and  
(e)Californium?

Laughing gas; Dmitri  
Mendeleev; Antifreeze;  
Linus Pauling (the only  
person to be awarded  
two unshared Nobel  
Prizes) Vitamin C;  
Copper and zinc;  
Copper and tin; Lenz's  
Law; J; Tungsten;



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you should be able to  
explain why the sky is  
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able to answer where  
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5. b) A fruit - a pomelo  
is the largest fruit in the  
citrus family. ` 6. An  
insulator. 7. The tongue.  
8. Ascepius. 9. The ...

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Exercise 1 - using moles  
and reacting masses ...

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complex calculations

Answers to 1.2

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*1.2 Amount Of*  
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Past Papers March,  
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to calculate reacting  
masses and product  
masses. Example 1 –  
reactant mass Question.  
When 12 g of carbon is  
burned in air, 44 g of  
carbon dioxide is  
produced.

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