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questions and exam-style questions on the 'Atoms, elements and compounds' and 'Mixtures' subtopics of the GCSE AQA Chemistry 'Atomic structure and the Periodic table' topic.

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Atoms are the fundamental building blocks of matter and are built from protons, neutrons and electrons. A useful way to approach the teaching of atomic structure is to consider the changing models proposed by scientists throughout history. The image here could represent a progression in how students understand atomic theory, starting with a simple billiard ball model at Key Stage 2/3 and ending with the electron cloud model at A Level.

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What is here? The periodic table below is based on the ones used by the different examination boards.. The group numbers 1 to 0 (the top ones) are used in most GCSE courses. The group numbers 1 to 18 were recommended by IUPAC in 1988. At the moment these are only used in OCR courses.

~~Periodic Table | Creative Chemistry~~

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What are A Levels? A Levels are examinations taken by students at around the age of 18. A Level Chemistry courses normally take two years to complete, with all the examinations taken at the end in the summer.

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Atomic Structure and Bonding Covers basic atomic properties (electronic structures, ionisation energies, electron affinities, atomic and ionic radii, and the atomic hydrogen emission spectrum), bonding (including intermolecular bonding) and structures (ionic, molecular, giant covalent and metallic).

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Efficient and creative employees translate to more motivation, happiness, profit and client satisfaction. Your organizational structure should be designed in such a way that it empowers creativity and efficiency. Employees' functions should be well-defined and they should know to whom they report.

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