

## Half Life Penny Lab Answers

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half life of an element, the amount of the radioactive element left and the amount that has decayed (the amount of the new element) we can figure out approximately how old a fossil or a sample of rock is. Materials: Each group of students needs: 100 pennies box with lid (shirt or shoe box works well) Procedure: 1. Answer the pre lab questions. 2.

~~Pennies Radioactive Half Life Lab~~

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~~Isotopic Pennies Lab Answers~~

The rate of decay is a fixed rate called a half-life. The half-life of a radioactive isotope refers to the amount of time required for half of a quantity of a radioactive isotope to decay. Carbon-14 has a half-life of 5730 years, which means that if you take one gram of carbon-14, half of it will decay in 5730 years. Different isotopes have different half-lives.

~~Half Life : Paper, M&M's, Pennies, or Puzzle Pieces - ANS~~

Half-life is the time it takes for half of a sample of an element to decay. For example, I have eight Francium-223 atoms. Francium-223 has a half life of about 22 minutes. This means that in 22...

~~Please help me with this half life lab? | Yahoo Answers~~

For every half-life interval of approximately  $1.25 \times 10^9$  years, approximately half of the individual K40 atoms in some population (a mineral sample) decay spontaneously to argon-40. Following...

~~10 points, half life penny lab questions for chemistry ...~~

The half-life of Technetium 99m is 6.0 h. (f) 12 mg ( $12 \times 10^{-3}$  g) of Technetium 99m is injected into a patient and starts to decay into Technetium 99. Calculate the amount of Technetium 99 present in the patient after 24 hours. 24 hours is 4 half-lives. The Technetium 99m reduces to 0.75 mg (or  $0.75 \times 10^{-3}$  g or  $7.5 \times 10^{-4}$  g or 0.0075).

~~ATOMS: HALF LIFE QUESTIONS AND ANSWERS~~

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Carbon-14 has a half-life of 5730 years, which means that if you take one gram of carbon-14, half of it will decay in 5730 years. Different isotopes have different half-lives. Half-Life : Paper, M&M's, Pennies, or Puzzle Pieces - ANS counts per second from a sample of iodine-131. The half life of iodine-131 is 8 days.

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- 1) Determine the average number of atoms remaining (not decayed) at each three-second time interval by adding the results from the two trials and dividing by two.
- 2) Create a table that compares time to the average number of atoms remaining at each time interval.

~~8.01 Half Life and Radioactive Decay: Half Life lab by ...~~

The time it takes for half of the remaining pennies to be removed is called the half-life. The half-life of the pennies in this model is about one toss. If you're using painted wooden cubes, the probability that a cube will land red side up is  $1/6$ . (Each cube has six sides, and only one of those sides is painted red.) It takes three tosses for about half the cubes to be removed, so the half-life of the cubes is about three tosses.

~~Radioactive Decay Model: Math and Chemistry Science ...~~

In this activity, students will learn the concept of half-life and how it relates to radioactive material. Students will determine, with a hands-on experiment, the half-life of a radioactive element, "Coinheadsium". Students will create and be able to recognize a graph representing the half-life of a radioactive element.

~~Understanding Half Life : Simulating the process of a ...~~

How does this graph compare to one where you shake the bag for 20 seconds, if in both cases you start out with 100 coins? What does this mean about materials with a longer half-life?

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