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Answer Ethanol is polar; ethane is not. The ethanol $-\text{OH}$ group can hydrogen-bond with water. 2. Calculation of pH from Hydrogen Ion Concentration What is the pH of a solution that has an H^+ concentration of (a) $1.75 \times 10^{-5} \text{ mol/L}$; (b) $6.50 \times 10^{-10} \text{ mol/L}$; (c) $1.0 \times 10^{-4} \text{ mol/L}$;

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