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relative number of atoms of each kind  
in a chemical compound. Reveals the

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number of atoms of each element contained in a single molecule ...

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## CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER

Answer the following questions in the space provided. 1. Label each of the following statements as True or False:

True a. If the formula mass of one molecule is  $x$  amu, the molar mass is  $x$  g/mol. False b. Samples of equal numbers of moles of two different chemicals

## 7 Chemical Formulas and Chemical Compounds

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Answers

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Quiz Modern Chemistry 33 . Name  
Section Quiz, continued Class Date 6.  
The Lewis structure of which of the  
following compounds does not follow  
the octet rule? a.  $\text{NH}_3$  b.  $\text{BF}_3$  c. d.  $\text{H}_2\text{O}$

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7. In a double covalent bond, a. one atom has more than eight valence electrons.

Home - David Brearley High School  
CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER  
Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law  
Modern Chemistry 137 Chapter Test  
Name Class Date Chapter Test A,  
continued \_\_\_\_\_ 19. In the figure on

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the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. \_\_\_\_\_20. In the figure on the previous page, the volume of titration standard

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