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CHAPTER 1 Matter and Change
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ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope. Atomic number is the total number of protons in the nucleus of each atom of an element. 2.

3 Atoms: The Building Blocks of Matter

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1 Matter and Change – HUBBARD'S CHEMISTRY

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Answer Guide Chapter 5-21

Answer Key

SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches per day in a prescribed area of the rain forest. The

2 Measurements and Calculations

phase energy (E_{ph}) -energy stored in system due to arrangement of particles that exert attractions on

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Answers Chapter 9
one another. (energy account involved when phase changes occur) -Attractions result in decrease of energy of a system of particles. -As particles become more tightly bound, their phase energy is lowered. -Solids possess lowest phase energy, and liquids possess more because the particles move freer.

Energy Study Guide

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ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the same or different elements is the .atom 3.

3 Atoms: The Building Blocks of Matter

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the constant motion of
electrons and the creation
of instantaneous dipoles;
forces act between all atoms
AND molecules; forces are
NOT present in
intramolecular forces;
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or molar mass.

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