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~~Exponential Equations: Half-Life Applications Why some  
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 ~~$N=N_0e^{-\lambda t}$  (Natural Log)~~~~

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Identify particles in transmutation equations, complete reactions to obey the conservation of mass and nuclear charge. Information A half-life ( $t_{1/2}$ ) is the time required for one-half of the nuclei of a radioisotope sample to decay to products. After each half-life, half of the existing radioactive atoms (parent element) have decayed into atoms of a new element (daughter element). Equation  $N = N_0(0.693)^{-t/t_{1/2}}$

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The time required for half of a sample of a radioactive  
isotopes to decay is called the half- life ( $t_{1/2}$ ). Critical Thinking  
Questions: 10.

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chemistry with a discussion of the four fundamental forces of physics (gravity, electro-magnetism, strong nuclear force, and weak nuclear force). The text covers the forces at work in an atomic nucleus and how they can lead to a balanced or unbalanced situation.

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The half-life of strontium-90 is 25 years. How much strontium-90 will remain after 100 years if the initial amount is 4.0 g? If the mass of a radioactive substance is 8 grams and it has half life of 4 hours, how much mass remains after 8 hours? Phosphorus-32 is radioactive and has a half life of 14.3 days.

Nuclear Half-Life - Chemistry | Socratic

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