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, Unit 2 | Phenol
| Resorcinol | Cresol
| Structure and uses,
hindi **Organic**

Chemistry Unit 2 Structure

Organic chemistry is generally considered to be the chemistry of the compounds of carbon with the exceptions of the carbonates and the simple oxides of

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Carbon. Around the beginning of the nineteenth century, those substances that were derived from plant and animal sources were described as organic and those from non-living sources were described as inorganic.

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coverage of the principles of organic chemistry. These principles are used to develop aspects of structure, bonding and isomers, together with organic reaction mechanism and functional group chemistry.

Summary of Unit 2: Organic Chemistry

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Alkanes

Kekule formulas is just organic chemistry's term for Lewis structures you have previously encountered. In condensed structural formulas, the bonds to each carbon are omitted, but each distinct structural unit (group) is written with subscript numbers

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designating multiple substituents, including the hydrogens.

1.12: Drawing Chemical Structures - Chemistry LibreTexts

The science of carbon-based chemistry is called organic chemistry, and it is the focus of this unit. You learned the

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basics of organic chemistry in your grade 11 chemistry course and you will review and extend this knowledge in this unit. The various types of organic compounds and their practical uses such as polymers and plastics will be ...

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Structure and Bonding. Lewis Structures in Organic Chemistry; Valency and Formal Charges in Organic Chemistry; sp^3 , sp^2 , and sp Hybridization in Organic Chemistry with Practice Problems; How to Quickly Determine The sp^3 , sp^2 , and sp

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Lengths and Bond
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More organic chemistry.... Instead, we show the structure of its repeating unit, the part that is repeated many times. To deduce the structure of a polymer from the monomer:

**Addition
polymerisation -
More organic
chemistry - AQA ...**

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11 SUMMARY AND
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and Physical
Properties of Alcohols
CA11i.i alkanols –
structure; CA11i.ii
alkanols – physical
properties (including

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those due to
intermolecular
hydrogen bonding);
ISA13 organic and
inorganic compounds
(alkanols (methanol,
ethanol, propanol) -

UNIT 11 FURTHER ORGANIC CHEMISTRY UNIT 11

Unit 2: Organic
Chemistry Unit code:
R/601/0352 Level: 4

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Credit value: 15 Aim

This unit develops the principles and practical techniques of organic chemistry.

Rationalisation of structure and bonding is used to aid understanding of reaction mechanisms and functional group conversions. Unit abstract

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Chemistry Unit 2 is a
linear qualification.

This means that
students will sit their
exams at the end of
the course. Various
topics are covered
including atomic
structure and the
periodic table,

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quantitative chemistry,
organic chemistry and
much more. It is
assessed by two
exams.

AQA GCSE Chemistry (Separate Science) Unit 2: Bonding ...

UNIT 7 –
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CHEMISTRY 2 UNIT

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7 SUMMARY AND
SYLLABUS
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Classification of
Organic Compounds
CA1 1ai classification
(broad classification
into straight chain,
branched chain and
alicyclic compounds),
CA1 1aii classification
– functional groups,

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CAei alkanes
(structure), CAfi
alkenes (structure),
CA11g

Alkanes

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4.11 Structure Determination - A- Level Chemistry

CO₃²⁻? Like ozone,
the electronic
structure of the
carbonate ion cannot

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be described by a single Lewis electron structure. Unlike O_3 , though, the Lewis structures describing CO_3^{2-} has three equivalent representations. 1. Because carbon is the least electronegative element, we place it in the central position:

2.4: Resonance -

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of carbon-based chemistry is called organic chemistry, and it is the focus of this unit. You learned the basics of organic chemistry in your grade 11 chemistry course and you will review and extend this knowledge in this unit.

Organic Chemistry

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AFM image of 1,5,9-tri
oxo-13-azatriangulen
e and its chemical
structure. A molecule
is an electrically
neutral group of two
or more atoms held
together by chemical
bonds. Molecules are
distinguished from
ions by their lack of
electrical charge.. In

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quantum physics,
organic chemistry,
and biochemistry, the
distinction from ions is
dropped and molecule
is often used when
referring to polyatomic
ions.

Molecule - Wikipedia

An ion (/ ? a? ? n,-?n
/) is an atom or
molecule that has a
net electrical

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charge. Since the charge of the electron (considered negative by convention) is equal and opposite to that of the proton (considered positive by convention), the net charge of an ion is non-zero due to its total number of electrons being unequal to its total number of protons. A

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Organic is a positively
charged ion ...

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