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content questions.

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energy. The energy an

object has because of its
motion. kinetic theory.

All matter consists of
tiny particles that are in
constant motion.

molecules/atom. the
particles in a gas usually
consist of. insignificant.

The particles in a gas

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are considered to be
small, hard spheres with
an _____ volume.
empty space ...

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Answers
All matter consists of
molecules/atom. the

particles in a gas usually
consist of. empty space.

what is between

particles in a gas? rapid,
constant, and random.

the motion of the
particles in a ...

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6. alkali metals. alkaline
earth metals. halogen.

cation. any metal in

Group 1A of the

periodic table. any metal

in Group 2A of the

periodic table. a

nonmetal in Group 7A

of the periodic table.

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Answers

any atom or group of
atoms with a positive
charge . alkali metals.
any metal in Group 1A
of the periodic table ...

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molecules/atom.

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constant motⁿ the

particles in a gas usually

consist of ...

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describe the
assumptions of kinetic
theory as applied to
gases-The particles in a
gas are considered to be
small, hard speres with
an insignificant volume-
The motion of the
paricles in a fas is rapid,

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Atmospheric Pressure.
Barometer. Two or
more different
molecular forms of the

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Answers

same element in the same physical state. A solid without an ordered internal structure. The result from the collisions of atoms and molecules in air with a device that is used to measure atmospheric pressure. Allotrope. Two or more ...

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energy. kinetic theory.
molecules/atom.

Answers. The

energy an object has
because of its motion.

All matter consists of
tiny particles that are in
constant motion. The
particles in a gas usually
consist of. The particles
in a gas are considered
to be small, hard sphere.

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kinetic energy. The
energy an object has
because of its motion.
kinetic ...

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Amorphous Solid.

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Essment
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STUDY.PLAY. Kinetic

Answers
Energy. The energy an object has because of its motion. Kinetic Theory .

The theory that states that all matter consists of tiny particles that are in constant motion.

When applied to gases makes the following

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fundamental
assumptions: The
particles in a gas are
small hard spheres with
insignificant volume,
The motion of gas
particles is ...

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and what to complete
considering mostly your
friends love reading?

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don't have such hobby?

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you to begin having that
hobby. You know,

reading is not the force.

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reading will guide you

to belong to in greater

than before concept of

life. Reading will be a

clear protest to reach

every ...

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essential lessons
associated with water
and aqueous systems.

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Aqueous ...

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of valence ... D^1 Pearson

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13. Atoms acquire the

stable electron structure

of a noble gas by losing

electrons. ... Answer the

following in the ...

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