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your score will appear
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<= => Which factor would not
affect the solubility of
sugar in water? ? Add salt
to the water ? Heat the
water and sugar ?

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have made about factors that affect solubility. For example, perhaps they have seen that vinegar and oil

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_____ grams of _____ that
will _____ in 100 g of _____
at _____

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section, you will explain

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Some important properties of water explain solution formation in terms of intermolecular forces between polar, ionic, and non-polar substances describe the relationship between solubility and

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temperature for solids,
liquids, and gases
communicate your under-

8.2 Factors That Affect Rate
of Dissolving and Solubility
Solubility is the quantity

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of solute (solid, liquid or gas) that will dissolve in a given amount of solvent (solid, liquid or gas) at definite temperature and pressure. Solubility depends on the physical and chemical property of solvent, solute

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and concentration of the solution. Want to see the full answer? Check out a sample textbook solution.

Using Table 8-2, indicate whether each of the

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non-8-en-2-one, belongs to the class of organic compounds known as ketones. These are organic compounds in which a carbonyl group is bonded to two carbon atoms $R_2C=O$ (neither R may be a hydrogen

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atom). Ketones that have one or more alpha-hydrogen atoms undergo keto-enol tautomerization, the tautomer being an enol.

8-Nonen-2-one | C₉H₁₆O -

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To measure concentration,
you compare the amount of
solute to the total amount
of solution. Why is
solubility useful in

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Identifying substances? You can identify a substance by its solubility because it is a characteristic property of matter. What

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Solubility vs. Temperature
for Solids H₂O) KI KCl 20
10 30 40 50 60 70 80 90 110
120 130 140 100 NaNO₃ KNO₃
HCl NH₄Cl NH₃ NaCl KClO₃
SO₂ Graph that shows the
dependence of solubility on

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temperature gases solids
saturated: solution has
"just right" amt. of solute;
on line unsaturated:
solution could

maximum grams of solute that

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will temperature Varies with

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Compounds in g 100 g of
Water 1 00oc 104 56 3 245
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