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Since 20 moles  $\text{NO}_2$  are produced, and it takes 3 moles  $\text{O}_2$  to produce 4 moles of  $\text{NO}_2$ , the answer would be 15 moles.  $15 \text{ moles} = 20 \text{ moles NO}_2 \times (3 \text{ moles O}_2 / 4 \text{ moles NO}_2)$ --- This is the molar ratio between the two.

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